

## Chemical reaction lab

### Materials:



2 Well plates

### Procedure:

1. Obtain 2 well plates
2. Place 6 drops of each chemical into the corresponding well.
3. Record your observations in your observation table.

\*Hold the well plate up and look in the bottom of each well for a precipitate.

\*IF there was no reaction write NR.

### Observation table:

***\*When you get to your station and set up the well plates number the top corner of each box in your observation table.***

	↓ $K_3PO_4$	↓ $KOH$	↓ $KI$	↓ $K_2C_2O_4$
$Co(NO_3)_2$ →				
$Pb(NO_3)_2$ →				
$CuSO_4$ →				
$Fe(NO_3)_3$ →				

### Follow up questions:

1. Write a balanced chemical equation for the following compounds:



2. What type of reaction do you think occurred where a precipitate formed?