Den	sity	of	sol	lids	S
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Name: \_\_\_\_\_

Question: Which cube will have the greatest density?

**Data Table:** 

Cube #	Mass (g)	Volume (cm³)	Density (g/cm³)	Identity of cube
1	14.0 g	15.6 cm <sup>3</sup>		
2	143.4 g	15.6 cm <sup>3</sup>		
3	138.8 g	15.6 cm <sup>3</sup>		
4	127.4 g	15.6 cm <sup>3</sup>		
5	43.5 g	15.6 cm <sup>3</sup>		
6	19.3 g	13.8 cm <sup>3</sup>		
7	17.8 g	13.8 cm <sup>3</sup>		
8	12.5 g	15.6 cm <sup>3</sup>		
9	6.0 g	14.4 cm <sup>3</sup>		

Conclusion: Least dense to most dense (use numbers):

Identify each of the cubes using the information below:

Material	Density
Brass	8.9 g/cm³
Aluminum	2.8 g/cm <sup>3</sup>
Copper	9.2 g/cm³
Oak	0.6-0.9 g/cm <sup>3</sup>
Poplar	0.3-0.5 g/cm <sup>3</sup>
PVC	1.4 g/cm <sup>3</sup>
Acrylic	1.3 g/cm <sup>3</sup>
Steel	8.2 g/cm <sup>3</sup>
Nylon	0.9 g/cm³